

APPLICATIONS OF QUANTUM MECHANICS IN DRUG DESIGN-

- A Brief Review

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ABSTRACT

The pharmaceutical industry is progressively operating in an era where development costs are constantly under pressure, higher percentages of drugs are demanded, and drug discovery process is a trial-and-error run. The application of quantum mechanical (QM) methods in drug discovery is becoming increasingly popular. This is a consequence of improvements in computing hardware, which have led to advances in QM-algorithm development and new applications of QM methods. The first-principles nature of quantum mechanics requires no implicit parameterization, and therefore should allow for the calculation of highly accurate molecular geometries and properties. However, in reality this level of accuracy needs to be balanced with the high computational expense of precise calculations as well as the rigorous pace of drug-discovery research. As a result, a number of approximations are required, resulting in numerous methodological developments in QM. This has spurred the development of QM methods for many computer-aided drug-discovery problems, such as describing molecular interactions, providing estimates of binding affinities, determining ligand energies, refining molecular geometries, scoring of docked protein–ligand poses, describing molecular similarity, and deriving descriptors for quantitative structure–activity relationships. This article shows importance of concept and applications of quantum mechanics in drug design and pharmaceutical field.

Key word: Quantum Mechanics (QM), Computer Aided Drug Design (CADD), QSAR.

Introduction:

Drug discovery plays an important role in the growth of any pharmaceutical company and society, as newer and safer drugs are launched in the market with the sole objective of improving the therapeutic value and safety of drugs. The pharmaceutical industry has consistently shown that it can discover and develop innovative medicines for a wide range of diseases.

Quantum mechanics is the branch of mechanics that deals with the mathematical description of motion and interaction of subatomic particles, incorporating the concepts of quantization of energy, wave particle duality, and correspondence principle.

Quantum mechanics is a technique, developed by Erwin Schrodinger (of the university of Zurich) in 1926. It is an important technique for “Drug design”. Erwin Schrodinger worked out mathematical expressions to describe the motion of an electron in terms of its energy. Those mathematical expressions are called wave equations, since they are based upon the concept that electrons show properties not only of particles but also of waves.

A wave equation has a series of solutions, called wave functions, each corresponding to a different energy level for the electron. For all but the simplest of systems, doing the mathematics is so time consuming that at present and super high-speed computers will someday change these only solutions can be obtained.

The QM method treats molecules as collections of nuclei and electrons without any reference to “chemical bonds”. QM is important in understanding the behaviour of systems at the atomic level. QM methods apply the laws of QM to approximate the wave function and to solve the Schrödinger equation. However, the Schrödinger equation cannot actually be solved for any but a one-electron system (the hydrogen atom), and approximations need to be made. According to QM, an electron bound to an atom cannot possess any arbitrary energy or occupy any position in space.

QM models are the most accurate, but also the most expensive methods in terms of time and computational resources, and are thus applied on small systems. CADD is aimed at improving the development and efficacy of drugs using modern computational tools that are fast and cost-effective compared to conventional methods. A biomolecular system can be simulated using molecular mechanics (MM), QM, or a hybrid method (QM/MM), depending on the research problem to be answered.

Objectives

The application of QM-based approaches in guiding SBDD is not new. QM has featured in some medically relevant chemistry calculations in providing informative descriptors for QSAR and 3-D conformation for ligands. QM methods offer the ability to provide an accurate representation of ligands and proteins where MM parameterization struggles. QM approaches hold promise in addressing pharmacological problems on the time scale demanded by drug-discovery research. After ups and downs in the perception of CADD and perhaps some overhyping of its promises in drug development, it could be said that CADD is becoming a routinely used component of drug discovery.

The use of QM and QM-MM approaches in computation of protein–ligand binding affinities has met with mixed success. However, the QM-MM approach appears to be of most benefit for low-resolution X-ray structures, where an incorrectly assigned ligand structure due to its MM force field is more likely. Studies demonstrate that the use of accurate charges, in many cases, leads to improvement in docking accuracy in a wide range of Protein Data Bank complexes.

QM should be applied to “lead” compounds to provide insight into the free-energy landscape. Most importantly, before embarking on CADD, it is appropriate to evaluate the diversity and demand of accuracy of molecules to be designed in the project.

Finally, QM methods have proved valuable in quantitative analysis of the energetics of ligand deformation on binding.

Methodology:

QM-MM MD

The hybrid QM-MM method is a molecular simulation approach that combines the accuracy of QM to treat the region of the system where the chemical process takes place and the speed of MM to the rest of the system, thus allowing for the study of chemical processes in large systems. This approach has been applied to target proteins, such as human acetylcholinesterase, heme peroxidases, metallo- β -lactamases, α -synuclein, ligase ribozymes, and trypsin.

QM-MM DOCKING

The accuracy of electric charges plays an important role in protein–ligand docking, which is why QM-MM calculations are incorporated into docking procedures. Fixed charges of ligands obtained from force-field parameterization are replaced by QM-MM calculations in the protein–ligand complex, treating only the ligand as the quantum region. This approach has been applied to target proteins, including F-actin, protein kinases, and metalloproteins.

QM-QSAR

QSAR models combined with QM-MM allow the prediction of drug ADMET and give reliable information on how the modification of a compound affects or improves pharmacokinetic/ pharmacological profiles. QM-QSAR has been applied to such targets as ACE and cytochrome P450.

QM-VS

The QM-VS approach provides unprecedented accuracy in structure-based binding-energy calculations that enable application of QM methodologies to noncovalent interactions in systems as large as protein–ligand complexes and conformational ensembles. This method bridges the gap between the high accuracy of QM and high-volume computations (VS) in drug research and has been applied to such targets as HIV1 integrase and butyrylcholinesterase.

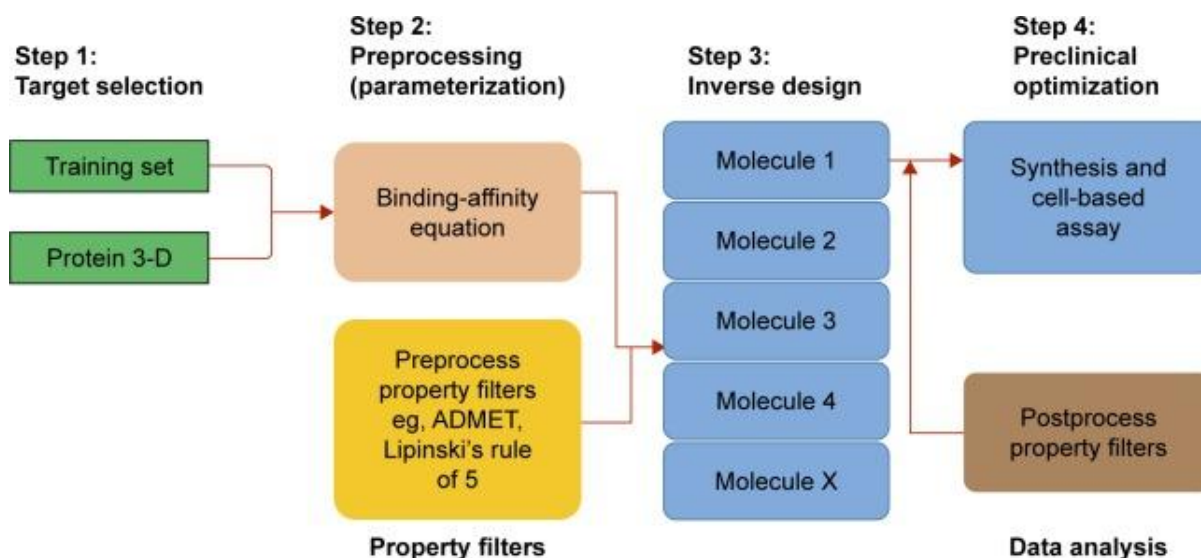


Fig.1 Implementation of QM in Pharmaceutical Companies Drug Design workflow

Recent QM developments in drug discovery:

Tangible advances in the use of QM to solve relevant pharmaceutical problems have been seen in the last decade, eg, the use of the hybrid QM-MM approach to determine the free-energy landscape of the enzymatic reaction mechanism.

The next step in the evolution of drug discovery is the routine use of QM in all levels of in silico Drug Design. Silico Based Drug Design is an important factor in the drug-discovery process, and designs more potent molecules with few alterations made, ie, derivatives of “lead” molecules.

In silico tools can be used to design molecules to investigate existing protein–ligand interactions, as well as explore the active site for any supplementary hydrophilic or hydrophobic interactions that can increase binding affinity. The use of in silico tools allows the testing of a theory in a short time frame, using high-throughput empirical methods.

The in-silico approach is fast and environmentally friendly, but it does not replace experimentation. Regardless, failures encountered in the pharmaceutical industry at the drug-discovery stage can be attributed to a number of factors that are not limited to wrong force-field parameters, especially for metals, disregard for protein flexibility or domain of applicability, or nonrigorous validation of the QSAR model.

QM, a method used to replicate an experimental work accurately, proffers a potential solution to the failures mentioned. Increasingly, QM-MM methods are being applied to enzymes that are drug targets, often with the aim of providing information for DD. Examples include the HIV1-replication enzymes: reverse transcriptase, protease, and integrase. The reaction mechanisms of these enzymes have been studied using the QM-MM approach. Other examples are G-protein-coupled receptors,

5-HT₄ receptors, design and evaluation of a novel class of FKBP12 ligands, and novel inhibitors of human DHFR.

Another development in DD research is the hybrid QM-MM method, developed to improve the accuracy of biomolecular simulations, QM docking, QM virtual screening, and QM-QSAR

Significance:

The application of QM-based approaches in guiding SBDD is not new. QM has featured in some medicinally relevant chemistry calculations in providing informative descriptors for QSAR and 3-D conformation for ligands. QM methods offer the ability to provide an accurate representation of ligands and proteins where MM parameterization struggles. QM approaches hold promise in addressing pharmacological problems on the time scale demanded by drug-discovery research. After ups and downs in the perception of CADD and perhaps some overhyping of its promises in drug development, it could be said that CADD is becoming a routinely used component of drug discovery.

Currently, sophisticated CADD tools are typically applied by modeling experts, but are increasingly spreading to the desktops of medicinal chemists as well. Ligand poses predicted from docking to receptors, such as metalloproteins, have been shown to resemble experiments more closely when partial charges are derived from QM or QM-MM calculations. The use of QM and QM-MM approaches in computation of protein–ligand binding affinities has met with mixed success. However, the QM-MM approach appears to be of most benefit for low-resolution X-ray structures, where an incorrectly assigned ligand structure due to its MM force field is more likely. Studies demonstrate that the use of accurate charges, in many cases, leads to improvement in docking accuracy in a wide range of Protein Data Bank complexes. The principal uncertainty at this point is whether this improved performance in docking can be noticed in other in silico methods.

Conclusion:

In this review, we have discussed how the implementation of QM-based methods could help the drug-discovery and DD process in the pharmaceutical industry. This review outlines the major roles played by QM in the DD workflow and its importance in the drug-discovery process to avoid “dead-end” lead compounds. This method could have strong impact in future drug development, because of the endless demand for new drugs and the short time frame pharmaceutical companies have in developing them. Pharmaceutical companies have to reach a compromise between accuracy and productivity by applying QM in their research. The selection of the most appropriate method (MM, QM, or QM-MM) during drug development is of extreme importance. QM should be applied to “lead” compounds to provide insight into the free-energy landscape. Most importantly, before embarking on CADD, it is appropriate to evaluate the diversity and demand of accuracy of molecules to be designed in the project, which in turn dictates the most appropriate approach to select. It is also possible to reparameterize approximate methods in order to improve the accuracy of results in specific reactions that require numerous energy evaluations. A number of studies have sought to incorporate QM and QM-MM into their approaches for calculating ligand–receptor binding affinities. These approaches show promising results, but require further development to be broadly applicable. Finally, QM methods have proved valuable in quantitative analysis of the energetics of ligand deformation on binding. Although computation of binding energies remain a challenging and evolving area, current QM approaches could offer detailed information on the nature and relative strengths of complex active-site interaction, which is valuable in molecular design. It is likely that QM will become a more prominent tool in the repertoire of the computational medicinal chemist. Therefore, modern QM approaches will play a more direct role in informing and streamlining the drug-discovery process. The insight gained from this review could serve as a cornerstone for medicinal chemists, industry R&D and clinicians. This could provide better understanding of the in silico tools in drug design and development with improved ADMET, pharmacokinetics and the timely assessment of property profiles.

References:

1. Drews J. Drug discovery: a historical perspective. *Science*. 2000;287:1960–1964. [[PubMed](#)] [[Google Scholar](#)]
2. Meldrum NU, Roughton FJ. Carbonic anhydrase: its preparation and properties. *J Physiol*. 1933; 80:113–142. [[PMC free article](#)] [[PubMed](#)] [[Google Scholar](#)]
3. Kuhlmann J. Drug research: from the idea to the product. *Int J Clin Pharmacol Ther*. 1997; 35:541–552. [[PubMed](#)] [[Google Scholar](#)]
4. Bharath EN, Manjula SN, Vijaychand A. In silico drug design: tool for overcoming the innovation deficit in the drug discovery process. *Int J Pharm Pharm Sci*. 2011; 3:8–12. [[Google Scholar](#)]
Parasuraman S. Toxicological screening. *J Pharmacol Pharmacother*. 2011; 2:74–79. [[PMC free article](#)] [[PubMed](#)] [[Google Scholar](#)]

5. Alomar MJ. Factors affecting the development of adverse drug reactions (review article) *Saudi Pharm J*. 2014; **22**:83–94. [[PMC free article](#)] [[PubMed](#)] [[Google Scholar](#)]
6. Jochum C, Gasteiger J. Canonical numbering and constitutional symmetry. *J Chem Inf Comput Sci*. 1977; **17**:113–117. [[Google Scholar](#)]
7. Balaban AT, Basak SC, Colburn T, Grunwald GD. Correlation between structure and normal boiling points of haloalkanes C1-C4 using neural networks. *J Chem Inf Comput Sci*. 1994; **34**:1118–1121. [[Google Scholar](#)]
8. Rucker C, Meringer M, Kerber A. QSPR using MolGen-QSPR: the example of haloalkane boiling points. *J Chem Inf Comput Sci*. 2004;**44**:2070–2076. [[PubMed](#)] [[Google Scholar](#)]
9. Augen J. The evolving role of information technology in the drug discovery process. *Drug Discov Today*. 2002; **7**:315–323. [[PubMed](#)] [[Google Scholar](#)]
10. Schertler GF. Overproduction of membrane proteins. *Curr Opin Struct Biol*. 1992; **2**:534–544. [[Google Scholar](#)]
11. Guide to protein purification. *Methods Enzymol*. 1990; **182**:1–818. No authors listed. [[PubMed](#)] [[Google Scholar](#)]
12. Pardi A. Isotope labelling for NMR studies of biomolecules. *Curr Opin Struct Biol*. 1992; **2**:832–835. [[Google Scholar](#)]
13. Pflugrath JW. Developments in X-ray detectors. *Curr Opin Struct Biol*. 1992; **2**:811–815. [[Google Scholar](#)]
14. Ealick SE, Walter RL. Synchrotron beamlines for macromolecular crystallography. *Curr Opin Struct Biol*. 1993; **3**:725–736. [[Google Scholar](#)]
15. Watenpaugh KD. Macromolecular crystallography at cryogenic temperatures. *Curr Opin Struct Biol*. 1991; **1**:1012–1015. [[Google Scholar](#)]
16. Pitman MR, Menz RI. Methods for protein homology modelling. *Bioinformatics*. 2006; **6**:37–59. [[Google Scholar](#)]
17. Taft C, Da Silva V, Da Silva C. Current topics in computer-aided drug design. *J Pharm Sci*. 2008; **97**:1089–1098. [[PubMed](#)] [[Google Scholar](#)]
18. Klebe G. Toward a more efficient handling of conformational flexibility in computer-assisted modelling of drug molecules. *Perspect Drug Discov Des*. 1995;**3**:85–105. [[Google Scholar](#)]
19. Kubinyi H. Combinatorial and computational approaches in structure-based drug design. *Curr Opin Drug Discov Devel*. 1998; **1**:16–27. [[PubMed](#)] [[Google Scholar](#)]
20. Mestres J, Rohrer DC, Maggiora GM. MIMIC – a molecular-field matching program: exploiting applicability of molecular similarity approaches. *J Comput Chem*. 1997; **18**:934–954. [[Google Scholar](#)]
21. Mandal S, Moudgil M, Mandal SK. Rational drug design. *Eur J Pharmacol*. 2009; **625**:90–100. [[PubMed](#)] [[Google Scholar](#)]
22. Hao GF, Yang GF, Zhan CG. Structure-based methods for predicting target mutation-induced drug resistance and rational drug design to overcome the problem. *Drug Discov Today*. 2012; **17**:1121–1126. [[PMC free article](#)] [[PubMed](#)] [[Google Scholar](#)]
23. Hedvat M, Emdad L, Das SK, et al. Selected approaches for rational drug design and high throughput screening to identify anti-cancer molecules. *Anticancer Agents Med Chem*. 2012; **12**:1143–1155. [[PMC free article](#)] [[PubMed](#)] [[Google Scholar](#)]
24. Szymkowski DE. Creating the next generation of protein therapeutics through rational drug design. *Curr Opin Drug Discov Devel*. 2005; **8**:590–600. [[PubMed](#)] [[Google Scholar](#)]
25. Druker BJ, Lydon NB. Lessons learned from the development of an Abl tyrosine kinase inhibitor for chronic myelogenous leukemia. *J Clin Invest*. 2000; **105**:3–7. [[PMC free article](#)] [[PubMed](#)] [[Google Scholar](#)]
26. Weisberg E, Manley PW, Breitenstein W, et al. Characterization of AMN107, a selective inhibitor of native and mutant Bcr-Abl. *Cancer Cell*. 2005; **7**:129–141. [[PubMed](#)] [[Google Scholar](#)]
27. Bazer FW, Spencer TE, Johnson GA. Interferons and uterine receptivity. *Semin Reprod Med*. 2009; **27**:90–102. [[PubMed](#)] [[Google Scholar](#)]
28. Gupta A, Mandal SK, Leblanc V, Descôteaux C, Asselin E, Bérubé G. Synthesis and cytotoxic activity of benzopyran-based platinum (II) complexes. *Bioorg Med Chem Lett*. 2008; **18**:3982–3987. [[PubMed](#)] [[Google Scholar](#)]
29. Mandal S, Davie JR. An integrated analysis of genes and pathways exhibiting metabolic differences between estrogen receptor positive breast cancer cells. *BMC Cancer*. 2007; **7**:181. [[PMC free article](#)] [[PubMed](#)] [[Google Scholar](#)]
30. Sharma PS, Sharma R, Tyagi T. Receptor tyrosine kinase inhibitors as potent weapons in war against cancers. *Curr Pharm Des*. 2009; **15**:758–776. [[PubMed](#)] [[Google Scholar](#)]
31. Grodsky N, Li Y, Bouzida D, et al. Structure of the catalytic domain of human protein kinase C beta II complexed with a bisindolylmaleimide inhibitor. *Biochemistry*. 2006; **45**:13970–13981. [[PubMed](#)] [[Google Scholar](#)]
32. Patwardhan B, Bodeker G. Ayurvedic genomics: establishing a genetic basis for mind-body typologies. *J Altern Complement Med*. 2008; **14**:571–576. [[PubMed](#)] [[Google Scholar](#)]
33. Atkins P, de Paula J. *Atkins' Physical Chemistry*. 8th ed. New York: Macmillan; 2006. [[Google Scholar](#)]
34. Rogers DW. *Computational Chemistry Using the PC*. 3rd ed. Hoboken (NJ): Wiley; 2003. [[Google Scholar](#)]
35. Lewars EG. *Computational Chemistry: Introduction to the Theory and Applications of Molecular and Quantum Mechanics*. 2nd ed. Alphen aan den Rijn, Netherlands: Kluwer; 2004. [[Google Scholar](#)]
36. Tully JC. Perspective on “Zur Quantentheorie der Molekeln” *Theor Chem Acc*. 2000; **103**:173–176. [[Google Scholar](#)]

37. Hotokka M. Calculation of vibrational frequencies by molecular mechanics. In: Chalmers JM, Griffiths PR, editors. *Handbook of Vibrational Spectroscopy*. Hoboken (NJ): Wiley; 2006. [[Google Scholar](#)]
38. Tannor DJ. *Introduction to Quantum Mechanics: A Time-Dependent Perspective*. Sausalito (CA): University Science Books; 2008. [[Google Scholar](#)]
39. Mortimer R. *Physical Chemistry*. 3rd ed. Cambridge (MA): Academic Press; 2008. [[Google Scholar](#)]
40. Lewars EG. *Computational Chemistry: Introduction to the Theory and Applications of Molecular and Quantum Mechanics*. Heidelberg: Springer; 2011. [[Google Scholar](#)]
41. Jensen F. *Introduction to Computational Chemistry*. 2nd ed. Hoboken (NJ): Wiley; 1999. [[Google Scholar](#)]
42. Perdew JP, Ruzsinszky A. Fourteen easy lessons in density functional theory. *Int J Quantum Chem*. 2010; **110**:2801–2807. [[Google Scholar](#)]
43. Burke KJ. Perspective on density functional theory. *J Chem Phys*. 2012; **136**:150901. [[PubMed](#)] [[Google Scholar](#)]
44. Grimme S, Antony J, Schwabe T, Mück-Lichtenfeld C. Density functional theory with dispersion corrections for supramolecular structures, aggregates, and complexes of (bio)organic molecules. *Org Biomol Chem*. 2007; **5**:741–758. [[PubMed](#)] [[Google Scholar](#)]
45. Hehre WJ. *A Guide to Molecular Mechanics and Quantum Chemical Calculations*. Irvine (CA): Wavefunction; 2003. [[Google Scholar](#)]
46. Dewar MJ, Zoebisch EG, Healy EF, Stewart JJ. AM1: a new general-purpose quantum mechanical molecular model. *J Am Chem Soc*. 1985; **107**:3902–3909. [[Google Scholar](#)]
47. Zerner M. *Reviews of Computational Chemistry*. Weinheim, Germany: Wiley-VCH; 1991. [[Google Scholar](#)]
48. Car R. Introduction to density-functional theory and ab initio molecular dynamics. *QSAR Comb Sci*. 2002; **21**:97–104. [[Google Scholar](#)]
49. Atkins PW, Friedman RS. *Molecular Quantum Mechanics*. 4th ed. Oxford: Oxford University Press; 2005. [[Google Scholar](#)]
50. Atkins PW, Friedman RS. *Molecular Quantum Mechanics*. 3rd ed. Oxford: Oxford University Press; 2005. [[Google Scholar](#)]
51. Levine IN. *Quantum Chemistry*. 5th ed. Upper Saddle River (NJ): Prentice Hall; 1999. [[Google Scholar](#)]
52. Roothaan CC. New developments in molecular orbital theory. *Rev Mod Phys*. 1951; **23**:69–89. [[Google Scholar](#)]
53. Hehre WJ, Radom L, Schleyer P, Pople J. *Ab Initio Molecular Orbital Theory*. Hoboken (NJ): Wiley; 1986. [[Google Scholar](#)]
54. Lewars EG. *Computational Chemistry*. Heidelberg: Springer; 2011. [[Google Scholar](#)]
55. Bao G, Suresh S. Cell and molecular mechanics of biological materials. *Nat Mater*. 2003; **2**:715–725. [[PubMed](#)] [[Google Scholar](#)]
56. Lipkowitz KB. Abuses of molecular mechanics: pitfalls to avoid. *J Chem Educ*. 1995; **72**:1070–1075. [[Google Scholar](#)]
57. Nagase S. Polyhedral compounds of the heavier group 14 elements: silicon, germanium, tin, and lead. *Acc Chem Res*. 1995; **28**:469–476. [[Google Scholar](#)]
58. Sella A, Basch H, Hoz S. Reactivity of strained compounds: is ground state destabilization the major cause for rate enhancement? *J Am Chem Soc*. 1996; **118**:416–420. [[Google Scholar](#)]
59. Balaji V, Michl J. New strained organic molecules: theory guides experiment. *Pure Appl Chem*. 1988; **60**:189–194. [[Google Scholar](#)]
60. Chakraborty S, Saha C. The Curtin-Hammett principle: a qualitative understanding. *Reson J Sci Educ*. 2016; **21**:151–171. [[Google Scholar](#)]
61. Morgan J, Greenberg A. Curtin-Hammett principle: application to benzene oxide-oxepin tautomers. *Struct Chem*. 2013; **24**:1945–1956. [[Google Scholar](#)]
62. Seeman JJ. The Curtin-Hammett principle and the Winstein-Holness equation: new definition and recent extensions to classical concepts. *J Chem Educ*. 1986; **63**:42–48. [[Google Scholar](#)]
63. Clark M, Cramer RD, Van Opdenbosch N. Validation of the general purpose Tripos 5.2 force field. *J Comput Chem*. 1989; **10**:982–1012. [[Google Scholar](#)]
64. Burkert U, Allinger NL. *Molecular Mechanics*. Washington: American Chemical Society; 1982. [[Google Scholar](#)]
65. Rappé AK, Casewit CJ. *Molecular Mechanics Across Chemistry*. Sausalito (CA): University Science Books; 1999. [[Google Scholar](#)]
66. Halgren TA. Merck molecular force field – I: basis, form, scope, parameterization, and performance of MMFF94. *J Comput Chem*. 1996; **17**:490–519. [[Google Scholar](#)]
67. Senn HM, Thiel W. QM/MM methods for biomolecular systems. *Angew Chem Int Ed Engl*. 1996; **48**:1198–1229. [[PubMed](#)] [[Google Scholar](#)]
68. Lin H, Truhlar DG. QM/MM: what have we learned, where are we, and where do we go from here? *Theor Chem Acc*. 2007; **117**:185–199. [[Google Scholar](#)]
69. Garcia-Vela A, Gerber RB. Hybrid quantum/semiclassical wave packet method for molecular dynamics: application to photolysis of Ar...HCl. *J Chem Phys*. 1993; **98**:427–436. [[Google Scholar](#)]

70. Sauer J, Sierka M. Combining quantum mechanics and interatomic potential functions in ab initio studies of extended systems. *J Comput Chem*. 2000; **21**:1470–1493. [[Google Scholar](#)]
71. Monard G, Prat-Resina X, Gonzalez-Lafont A, Lluch JM. Determination of enzymatic reaction pathways using QM/MM methods. *Int J Quantum Chem*. 2003; **93**:229–244. [[Google Scholar](#)]
72. Honarparvar B, Kruger HG, Maguire GE, Govender T, Soliman ME. Integrated approach to structure-based enzymatic drug design: molecular modeling, spectroscopy, and experimental bioactivity. *Chem Rev*. 2014;**114**(1):493–537. [[PubMed](#)] [[Google Scholar](#)]
73. Menikarachchi LC, Gascon JA. QM/MM approaches in medicinal chemistry research. *Curr Top Med Chem*. 2010; **10**:46–54. [[PubMed](#)] [[Google Scholar](#)]
74. Liao C, Nicklaus MC. Tautomerism and magnesium chelation of HIV-1 integrase inhibitors: a theoretical study. *ChemMedChem*. 2010; **5**:1053–1066. [[PMC free article](#)] [[PubMed](#)] [[Google Scholar](#)]
75. Cho AE, Guallar V, Berne BJ, Friesner R. Importance of accurate charges in molecular docking: quantum mechanical/molecular mechanical (QM/MM) approach. *J Comput Chem*. 2005; **26**:915–931. [[PMC free article](#)] [[PubMed](#)] [[Google Scholar](#)]
76. Zhou T, Huang D, Caflisch A. Is quantum mechanics necessary for predicting binding free energy? *J Med Chem*. 2008; **51**:4280–4288. [[PubMed](#)] [[Google Scholar](#)]
77. Cloudera The Hadoop ecosystem. 2016. [Accessed November 28, 2016]. Available from: <http://www.cloudera.com/training/library/hadoop-ecosystem.html>.
78. Chu WT, Wang J. Energy landscape topography reveals the underlying link between binding specificity and activity of enzymes. *Sci Rep*. 2016; **6**:27808. [[PMC free article](#)] [[PubMed](#)] [[Google Scholar](#)]
79. Ribeiro AJ, Santos-Martins D, Russo N, Ramos MJ, Fernandes PA. Enzymatic flexibility and reaction rate: a QM/MM study of HIV-1 protease. *ACS Catal*. 2015; **5**:5617–5626. [[Google Scholar](#)]
80. Greer J, Erickson JW, Baldwin JJ, Varney MD. Application of the 3-dimensional structures of protein target molecules in structure-based drug design. *J Med Chem*. 1994; **37**:1035–1054. [[PubMed](#)] [[Google Scholar](#)]
81. Kubinyi H. Structure-based design of enzyme inhibitors and receptor ligands. *Curr Opin Drug Discov Devel*. 1998; **1**:4–15. [[PubMed](#)] [[Google Scholar](#)]
82. Babine RE, Bender SL. Molecular recognition of protein-ligand complexes: applications to drug design. *Chem Rev*. 1997; **97**:1359–1472. [[PubMed](#)] [[Google Scholar](#)]
83. Anderson AC. The process of structure-based drug design. *Chem Biol*. 2003; **10**:787–797. [[PubMed](#)] [[Google Scholar](#)]
84. Marrone TJ, Briggs JM, McCammon JA. Structure-based drug design: computational advances. *Annu Rev Pharmacol Toxicol*. 1997; **37**:71–90. [[PubMed](#)] [[Google Scholar](#)]
85. Raha K, Peters MB, Wang B, et al. The role of quantum mechanics in structure-based drug design. *Drug Discov Today*. 2007; **12**:725–731. [[PubMed](#)] [[Google Scholar](#)]
86. Davis AM, Teague SJ, Kleywegt GJ. Application and limitations of X-ray crystallographic data in structure-based ligand and drug design. *Angew Chem Int Ed Engl*. 2003; **42**:2718–2736. [[PubMed](#)] [[Google Scholar](#)]
87. Tame JR. Scoring functions: a view from the bench. *J Comput Aided Mol Des*. 1999; **13**:99–108. [[PubMed](#)] [[Google Scholar](#)]
88. Warren GL, Andrews CW, Capelli AM, et al. A critical assessment of docking programs and scoring functions. *J Med Chem*. 2006; **49**:5912–5931. [[PubMed](#)] [[Google Scholar](#)]
89. Enyedy IJ, Egan WJ. Can we use docking and scoring for hit-to-lead optimization? *J Comput Aided Mol Des*. 2008; **22**:161–168. [[PubMed](#)] [[Google Scholar](#)]
90. Kim R, Skolnick J. Assessment of programs for ligand binding affinity prediction. *J Comput Chem*. 2008; **29**:1316–1331. [[PMC free article](#)] [[PubMed](#)] [[Google Scholar](#)]
91. Johnson SR. The trouble with QSAR (or How I learned to stop worrying and embrace fallacy) *J Chem Inf Model*. 2008; **48**:25–26. [[PubMed](#)] [[Google Scholar](#)]
92. Hawkins DM. The problem of overfitting. *J Chem Inf Comput Sci*. 2004; **44**:1–12. [[PubMed](#)] [[Google Scholar](#)]
93. Halgren TA, Damm W. Polarizable force fields. *Curr Opin Struct Biol*. 2001; **11**:236–242. [[PubMed](#)] [[Google Scholar](#)]
94. Teague SJ. Implications of protein flexibility for drug discovery. *Nat Rev Drug Discov*. 2003; **2**:527–541. [[PubMed](#)] [[Google Scholar](#)]
95. Davis AM, Teague SJ. Hydrogen bonding, hydrophobic interactions, and failure of the rigid receptor hypothesis. *Angew Chem Int Ed Engl*. 1999; **38**:737–749. [[PubMed](#)] [[Google Scholar](#)]